ERRATUM

ARTHUR J. BIRCH, B. M. RATNAYAKE BANDARA, KEVIN CHAMBERLAIN, BRIAN CHAUNCY, PETER DAHLER, ANTHONY I. DAY, IAN D. JENKINS, LAWRENCE F. KELLY, THONG-CHAK KHOR, GUY KRETSCHMER, ANDRIS J. LIEPA, ACHARAN S. NARULA, WARWICK D. RAVERTY, EZIO RIZZARDO, CHARLES SELL, G. RICHARD STEPHENSON, DAVID J. THOMPSON and DAVID H. WILLIAMSON: Organometallic compounds in organic synthesis—XI. The strategy of lateral control of reactivity: tricarbonylcyclohexadieneiron complexes and their organic synthetic equivalents. *Tetrahedron* 37, Suppl. 1, 289 (1981).

The Printer regrets an error on p. 301. The third and fourth paragraphs on the right hand side should read:

Steric hindrance of a classical reduction in the ergosterol series

Tricarbonyl(ergosta - 3,5,7,22 - tetraen - 3 - ol acetate)iron (37). To a soln of the ergosta-3,5,7,22-tetraen-3-ol-acetate⁶⁶ (0.872 g, 2 mmol) in anhydrous benzene (15 ml) was added Fe₃(CO)₁₂ (2.05 g, 4 mmol) and the mixture was stirred and heated at reflux under N₂ for 20 hr. After filtration through Celite and evaporation of the solvent, the residue was chromatographed on silica (30 g). Elution with benzene/petroleum ether (1:1) removed traces of unreacted Fe₃(CO)₁₂. The tricarbonyl (ergosta-3,5,7,22-tetraen-3-ol-acetate)iron (37) was eluted with benzene and recrystallised from EtOH to give bright yellow needles (0.96 g, 83%), m.p. 148–150°. IR: 2025, 1960, 1940, 1750, 1660 cm⁻¹. NMR: δ 2.07 (3 H, s); 4.96 (1 H, d, J = 4 Hz, 6-H or 7-H); 5.14 (3 H, m, 6-H or 7-H, 22-H, 23-H); 5.37 (1 H, d, J = 2 Hz, 4-H). UV: $\lambda_{max}253$ (ϵ 2200) and 207 nm (ε 27000). (Found: C, 69.2 H, 7.7. C₃₃H₄₄FeO₅ requires: C, 68.8 H, 7.7%).

Tricarbonyl(ergosta - 5,7,22 - trien - 3 - one)iron (38). A mixture of 37 (0.576 g, 1 mmol), KHCO₃ (0.2 g, 2 mmol) and MeOH (20 ml) was stirred and heated at reflux under N₂ for 0.5 hr. After cooling to room temp the product was collected, washed with water MeOH (1:1), dried in vacuo and recrystallised from EtOAc/MeOH to give bright yellow prisms of tri-carbonyl(ergosta-5,7,22-trien-3one)iron (0.49 g, 92%), m.p. 181-182° (in a tube sealed in vacuo) (lit.³⁹ 145-147° dec). On a Kofler block in air these crystals begin to darken at approximately 145° but do not melt. IR: 2020, 1945, 1710 cm⁻¹. NMR: δ 4.84 (1 H, d, J = 4 Hz, 6-H or 7-H); 5.16 (3 H, m, 6-H or 7-H and 22-H, 23-H). MS: 534 (M⁺). UV: 240 (18300) and 213 nm (ε 20200). No max at 253 nm in CHCl₃ [lit.³⁹ UV: (CHCl₃) 252 nm (£ 10800)]. (Found: C, 69.4 H, 7.8. Calc for C₁₁H₄₂FeO₄: C, 69.7 H, 7.9%).